

SMALLEST MATERIAL OBJECT

Invisible Particles with All the Individuality of Planets.

THEY ARE ALWAYS IN CEASELESS MOTION

No Minute that Millions Together Would Sincerely Make a Grain of Sand—How They Transmit Heat and Light.

(Copyright, 1896, by Sir Robert Ball.)

The man of science is at one time exploring the depths of space and becoming there conversant with magnitudes so vast as to tax his powers of conception to the utmost. At another time he is engaged in the study of objects far too minute to admit of their direct perception by any of his senses. That potent weapon for the investigation of nature which is supplied by the laws of mathematics is equally available for the discussion of the phenomena presented in such a mighty system as the Milky Way, or for tracing the movements of those atoms of matter so exceedingly small that they must necessarily elude every endeavor to perceive them.

It was at one time supposed that every substance must be susceptible of indefinite subdivision. If we took a material object, say for example a sheet of paper, and cut it in half, and repeated the operation again and yet again, ever halving one of the portions, which was left by the first cutting, it used to be thought that though the fragments of paper thus produced were ever growing less and less, yet it should be possible to continue this subdivision indefinitely, if only sufficient delicacy of manipulation were forthcoming. In other words, the matter to be divided would be no piece of paper so small but that it would admit of division again into two other pieces, each of which would be still the substance possessing the qualities of paper. But now we know the indefinite divisibility which is here postulated is not the property of matter as it is in nature. Only a finite number of divisions could be made of any material fragment which is arrived at—though perhaps itself veritable paper—could not receive any further subdivision, without ceasing to be paper. Of course, however, that in the case supposed we are operating upon a substance which is manifestly of a composite character. The result of the subdivision, when carried on sufficiently far, therefore, necessarily discloses the ultimate ingredients of which the composite material known as paper is formed. Let us therefore take the illustration of a substance which, as far as we can tell, is absolutely homogeneous, inasmuch as it consists only of a single element. I select for this purpose a piece of iron, and divide it into two portions. Let each portion be subdivided again, and yet again, until at last it shall have been reduced to the minutest particles. Each one of the little pieces so obtained will still possess all the qualities of iron. We shall further imagine that we are proceeding to subdivide the iron, carrying on the subdivision of an iron particle to a point much beyond that which any mechanical appliance at our disposal can effect. We will suppose that we are able to continue the subdivision of the iron long after the particles have become too minute to be visible, even in the most powerful microscope, and that we are, however, taught us that though this subdivision can be carried on so far, yet it cannot be protracted indefinitely. A point will be reached at last beyond which the little particles, though still possessing all the qualities of iron, would refuse to admit of any further subdivision. The particle in question may, no doubt, be composed of many atoms, but we could separate these parts by any means we please. This piece of iron which cannot be further reduced is called an atom. The derivation of this word indicates that the object to which it is applied is a something which cannot be cut. We are thus led to the conception that all matter on the earth or throughout the universe is constituted of aggregations of atoms. The sun itself is no more than an enormous mass of atoms, and the same is true of those ultimate atoms out of which all material objects must be composed.

LACK OF INFORMATION.

There is perhaps no other department of scientific research so generally strikingly deficient in our knowledge of nature. Any adequate information as to what these atoms of matter really are has been hitherto wanting. A few facts may be stated, however, at all events that the atoms are so minute that millions of them would be required to be put together to form the bulk of a small grain of sand. It would, however, be quite erroneous to suppose that these objects are so minute that their structure is therefore simple. This is by no means the case. Some phenomena prove unmistakably that the atoms are not simple objects, for instance as those of iron, which I have already used as an illustration, must be anything but simple objects. They should rather be regarded as consisting of highly complex character and as elaborately formed from many different portions, these portions being in many cases animated by rapid and intricate movement. Indeed, it would seem that the experience of the grosser objects which alone are perceptible to our senses would be capable of affording any adequate conception of the extraordinary liveliness of atoms. I must try to explain some of their varied activities.

Let us think of the steam in the cylinder of a steam engine. The steam presses upon the piston and thus it is that it can do its work. In our ordinary language we say that this work is done by the pressure of the steam on the piston, and everybody understands what is meant when we thus speak of the pressure of the steam. If, however, we look a little more closely into the matter, we shall find that what the engineer understands by the pressure of the steam, has to be regarded as a somewhat unexpected light when the ultimate constitution of steam is considered. The water from which the steam is made, of course, produced by the chemical union between the two gases, oxygen and hydrogen. Each molecule of steam is in fact the result of the union between two atoms of hydrogen and one of oxygen. Steam thus consists of molecules of which each is divided into lesser particles of the same substance, namely, water. If any subdivision of a molecule of steam were effected by the parts into which it would be separated would not be water, they would be the atoms of the constituent gases from which that water was made. The steam in the cylinder of the engine is to be regarded as consisting of a vast number of molecules of water. Each of these molecules is in a state of rapid motion. It is hurrying along with a speed which is sometimes slower and sometimes faster than that of a rifle bullet. Even in a very small portion of space the multitude of these molecules is prodigious. The number of them which are required to form such a mass of steam as atmospheric pressure would suffice to fill a lady's thimble is to be represented by many billions. As these molecules are in such close contiguity, and as they are in motion, it would be not surprising to find that collisions frequently take place between them. The effect of a collision will be to divert each of the impinging molecules from the path in which it was proceeding before the collision took place, so that it bounds off again in some other direction. This new direction is similarly pursued until the molecule is turned aside by the next collision. These operations take place so rapidly that each of the molecules will experience millions of collisions in each second.

MOLECULES OF STEAM.

As the molecules of steam in the cylinder dash about with their tremendous velocity they rain innumerable myriads of little blows upon the bottom of the piston. The effect of these impacts is to push the piston upward. Indeed what the engineer calls the pressure of the steam is merely the result of the myriads of little impulsive shocks which are given into which would be rapidly moving molecules. If the heat from the boiler is still applied, while the steam generated is not allowed to escape, then of course the pressure of the steam rises. But we may state that this means a different manner.

The increase of pressure arises from the fact that the temperature of the steam is increased. As the temperature increases the rate at which the molecules are moving also becomes greater. There is in fact a definite relation between the temperature of the steam and the average rate at which its molecules are moving. The more the temperature the more the speed; the less the temperature the less the speed. The increase of the pressure within the boiler is equivalent to an increase in temperature of the steam. This corresponds to an increase of the average speed with which the molecules are animated. But with increased velocities of the molecules there would be a corresponding increase in the violence of the blows which they administer to the inside of the boiler and consequently as we say the pressure in the boiler is augmented. In certain circumstances, however, these blows may become so numerous and so energetic that the tension of the iron or steel of which the boiler is constituted may no longer be able to withstand the strain to which it is exposed, in which case explosion will be the result. This illustration will at all events show to a certain extent how the temperature of a gas is connected with the average speed by which its molecules are animated.

ELEMENTS OF FUSION.

When two liquids, such as brandy and water, are placed together in a tumbler a complete fusion takes place. No doubt in this case the act of fusion is generally accompanied by the way in which one liquid is poured into the other. Fusion would, however, proceed without such assistance; it could not be prevented if the two liquids were brought into contact. Suppose the water has been placed first in the glass and the spirit being the lighter liquid has been carefully poured on top. There will be at first a marked difference between the two strata; a gradual blending of the two liquids, by which the two liquids will become one. Notwithstanding the relative lightness of the spirit, it cannot remain permanently distinct from the water as the molecules of oil do when mixed with the same circumstances. In due time the spirit descends through the water and the water ascends through the spirit, so that the two liquids will ultimately become completely as if the two liquids had been shaken together in a bottle. Thus we see that the spirit, though actually lighter than the water, gradually sinks through the water, though heavier than the spirit, gradually makes its way upward. The explanation of this phenomenon can be given in a way which will remember that each of the two liquids in question is made up of molecules in motion. Across the boundary which at first divides the two liquids, the molecules of one or the other of either liquid occasionally dashes, and by the incessant repetition of this process the blending is ultimately accomplished. The molecules have, therefore, the power of transferring part of the energy of their vibrations to the other, and this of originating waves of motion in the other. The molecules of the gas liquid, when in the liquid state, are not so much restricted as if they were in a gaseous state. Each molecule is free to move in any direction, and in the liquid state of freedom. In this case each molecule can only be detached from its association with the other molecules in order to be associated with another molecule. Such interchanges of alliance among the liquid molecules are, however, incessantly taking place, and it happens that the molecules of the spirit become gradually dispersed through the water, while on the other hand the molecules of the water gradually penetrate through the spirit, until at last the two fluids become completely blended.

ATOMS IN SOLIDS.

A solid substance, such as a piece of cold iron, may seem to us to be quite devoid of movement in its ultimate parts. We have, however, the best reasons for knowing that if we had organs of sense some millions of times more acute than those which nature has endowed us we should find that the molecules even of a piece of cold iron were animated by the liveliest movements. In fact, each molecule is composed of any body which may be termed solid, the movements of the molecules are of much more restricted character than they are in the case of a gas or even of a liquid. The extent of the movements of the particles of a solid are confined within very narrow limits. Each molecule, in fact, remains, generally speaking, in permanent association with the other molecules, and it is only in the most exceptional cases that it is disconnected. This is illustrated by the obvious truth that if a piece of solid copper and a piece of solid zinc are placed even in the closest contact no fusion of the two substances will take place. The movements of the molecules in the zinc are so narrowly restrained that they do not cross the boundary which separates them from the molecules of copper are also confined in their movements within the mass to which they originally belonged. If, however, these two metals are heated to a temperature at which they have been melted into a fluid state, then the two fluids, if placed in contact, will speedily diffuse one into the other. For under the influence of heat the amplitudes of the movements of the molecules have been so increased that they are now able to shake themselves free from their original attachments. The atoms of the zinc can thus cross the boundary which separates them from the atoms of the copper, and the homogeneous material known as brass is the result.

TEXTURE OF CRACKER BALLS.

Lord Kelvin has given a striking illustration to show the nature of the atoms of matter. Imagine that a rain drop the size of a pea were to be magnified into a globe as large as the earth. Let us suppose that each of the molecules in the drop of water were to be at the same time magnified in the like proportion, then we know that the increased size of the molecules would be so increased that they would be able to shake themselves free from their original attachments. The atoms of the zinc can thus cross the boundary which separates them from the atoms of the copper, and the homogeneous material known as brass is the result.

DECREASED EXPORTS.

While this criminally foolish law opened our ports to the scrawny cattle of Mexico, it has reduced our own cattle values to values down \$60,000,000, perhaps it increased our exportations? Perhaps this was a realization of that beautiful dream about the prospect of the American farmer to export his hides to the United States. The McKinley law helped the American farmer to export his hides to the United States. The McKinley law helped the American farmer to export his hides to the United States. The McKinley law helped the American farmer to export his hides to the United States.

THE HIDE BUSINESS.

Farmers understand that the hide business is important and, as a rule, goes up and down with the fluctuations of the cattle industry. In the year 1895 the hides from the surplus stock of 37,000,000 cattle and had to import \$26,000,000 worth of hides from the United States. The same time we exported \$1,250,000 worth of hides. Under reciprocity we increased our exports from \$1,225,895 in 1892 to \$3,972,494 in 1894, before the McKinley law was repealed. At the same time we reduced our imports from \$26,000,000 in 1892 to \$16,000,000 in 1894. The American farmer sold the difference between the calendar year 1895 and 1894, immediately after the passage of the Wilson bill, the exportation of hides fell to \$2,825,947 and the imports increased from \$16,000,000 to \$36,432,993.

THE DEAD HAND.

This is the withering effect of the dead hand of democracy felt in both agricultural and industrial endeavor, and in no industry is the lesson so clear as in the cattle and hide trade. I have shown that this legislation introduced uncertainty where certainty existed, brought about undervaluation and instigated fraud, increased importation to a point where it no longer became profitable for American farmers to raise and export such hides, and it decreased our exports and increased our imports. This is always the result of a democratic tariff. Under both tariffs hides were free.

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DELICACY OF FIBRES.

The thickness of a fiber of silk as wound on the cocoon is about one-five-hundredth of an inch. Prof. Boys has drawn fibers of quartz so fine that if 100 of them were twisted into a cable its thickness would be that of an ordinary piece of fiber or unspun silk. But this statement, remarkable as it may seem, is by no means adequate to express the highest order of fineness which is attainable in the drawing of the quartz filaments. They have indeed been drawn with such excessive delicacy that they can no longer be perceived by the naked eye. Indeed, Prof. Boys has

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If the facts which I propose to present do not prove the above assertion beyond a doubt I shall be glad to hear from any doubting Thomases on the subject. These data have not been gathered up haphazard, but carefully collected especially for this article by William F. Carroll, a statistical expert, acquired in his own right, who has visited personally the principal establishments of this city and talked with those engaged in the business.

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Decreased Exports and Increased Imports Under the Law Which Candidate Bryan Championed and Helped to Frame.

CHICAGO, Sept. 11.—(Correspondence of The Bee.)—I have just come from a long article from Chicago, showing the condition of its manufacturing industries, and it is proposed in this to take up an industry bearing more particularly on agriculture.

In common with all other business, the vast cattle interests of this city have suffered most severely by reason of the Wilson-Gorman law. While many of the troubles the people are suffering from may be attributed to "lack of confidence" and disturbance arising from the threatened free and unlimited export of silver, the trouble of the Chicago cattlemen is the old-fashioned complaint, for which the election of William McKinley is the best known specific remedy.

If the facts which I propose to present do not prove the above assertion beyond a doubt I shall be glad to hear from any doubting Thomases on the subject. These data have not been gathered up haphazard, but carefully collected especially for this article by William F. Carroll, a statistical expert, acquired in his own right, who has visited personally the principal establishments of this city and talked with those engaged in the business.